

Preliminary communication

Preparation of an optically active and other organosilicon derivatives of π -cyclopentadienyliron dicarbonyl

R.J.P. CORRIU and W.E. DOUGLAS

Laboratoire de Chimie des Organométalliques, Université des Sciences et Techniques du Languedoc, 34060 Montpellier-Cedex, France

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SUMMARY

The preparation and some properties of the compounds $(\pi\text{-Cp})\text{Fe}(\text{CO})_2\text{SiR}_3$ ($\text{R}_3 = \text{Me}_2\text{Ph}$, MePh_2 , Ph_3 , $\text{Ph}_2(\text{CH}=\text{CH}_2)$ or $\text{MePh}(1\text{-Np})$), optically active $(-)(\pi\text{-Cp})\text{Fe}(\text{CO})_2[\text{SiMePh}(1\text{-Np})]$ and $(\pi\text{-Cp})\text{Fe}(\text{CO})(\text{PPh}_3)\text{SiPh}_3$ are reported.

Compounds containing silicon to transition metal bonds are currently of interest¹, but few possessing an optically active silicon atom have been isolated^{2–4}.

The new compounds $(\pi\text{-Cp})\text{Fe}(\text{CO})_2\text{SiR}_3$ (I) ($\text{R}_3 = \text{Me}_2\text{Ph}$, MePh_2 , Ph_3 , $\text{Ph}_2(\text{CH}=\text{CH}_2)$ or $\text{MePh}(1\text{-Np})$) were obtained in 60–80% yield by reaction of $[(\pi\text{-Cp})\text{Fe}(\text{CO})_2]^- \text{Na}^+$ with R_3SiCl in THF. The complexes are yellow to orange crystalline solids and the solutions are air sensitive. No pure products were isolated from the reactions with R_3SiCl ($\text{R}_3 = (\text{OMe})\text{Ph}(1\text{-Np})$, $(\text{H})\text{Ph}(1\text{-Np})$, $(\text{CH}=\text{CH}_2)\text{Ph}(1\text{-Np})$ or $\text{Ph}_2(\text{CH}_2\text{CH}=\text{CH}_2)$), although in some cases oils were obtained which decomposed on attempted distillation or chromatography. It seems that the (1-Np) group has a destabilising effect on the compounds, sometimes precluding isolation. The compound $(\pi\text{-Cp})\text{Fe}(\text{CO})_2\text{SiMe}_2(\text{H})$ has previously been prepared⁵.

UV irradiation of a dilute solution of compound I ($\text{R}_3 = \text{Ph}_3$) in n-hexane in the presence of PPh_3 gave $(\pi\text{-Cp})\text{Fe}(\text{CO})(\text{PPh}_3)\text{SiPh}_3$ (II). However photolytic irradiation of complex I ($\text{R}_3 = \text{Ph}_2(\text{CH}=\text{CH}_2)$) resulted in decomposition. The compound $(\pi\text{-Cp})\text{Fe}(\text{CO})(\text{PPh}_3)\text{SiMe}_3$ has previously been isolated⁶.

Some chemistry of compound I ($\text{R}_3 = \text{Ph}_2(\text{CH}=\text{CH}_2)$) has been studied and is similar to that of $(\pi\text{-Cp})\text{Fe}(\text{CO})_2\text{SiMe}_3$ ⁷. Complex I did not react with the reagents EtO^- , NaBH_4 , KHF_2 , H_2O /dioxane or a solution of anhydrous HCl in pentane, however the Si–Fe bond is cleaved by chlorine to give $\text{Ph}_2(\text{CH}=\text{CH}_2)\text{SiCl}$. The ¹H NMR spectrum of a solution of I in CCl_4 shows a band of multiplicity 12 assignable to the vinyl group.

From (+)-MePh(1-Np)SiCl ($[\alpha]_D^{25} +6.3^\circ$; c 0.84 (cyclohexane)), analytically pure yellow crystals of (-)-(π -Cp)Fe(CO)₂SiMePh(1-Np) ($[\alpha]_D^{25} -25.5^\circ$; c 0.67 (cyclohexane); m.p. 108.5–110.5°) were obtained. The racemic compound melts at 124–126° (see Table 1). Studies are in progress on the optically active compound and its reactions.

All new compounds have been fully characterised by IR and ¹H NMR spectroscopy, analyses, and in some cases by mass spectrometry.

TABLE I

Compound	R _n	Colour	m.p. (°C)	Yield (%)	$[\alpha]_D^{25}$ (cyclohexane)
I	Me ₂ Ph	Yellow	68–68.5	72	1998, 1947
I	MePh ₂	Yellow	103.5–105.5	78	2002, 1951
I	Ph ₃	Yellow	164–165.5	60	2002, 1952
I	Ph ₂ (CH=CH ₂)	Yellow	114–114.5	68	2000, 1949
I	MePh(1-Np)	Yellow	124–126	64	2000, 1950
II		Orange	199 (dec.)	92	1916

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